Tetrahedron Letters No.3, pp. 281-284, 1967. Pergamon Press Ltd. Printed in Great Britain.

> 2-AMINO-2-THIAZOLINE. III.^{1,2} THE DIFFERING BEHAVIOR OF PHENYLISOTHIOCYANATE AND PHENYLISOCYANATE TOWARD 2-AMINO-2-THIAZOLINE

> > Daniel L. Klayman and James J. Maul

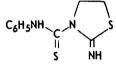
Division of Medicinal Chemistry, Walter Reed Army Institute of Research, Washington, D. C. 20012 U.S.A.

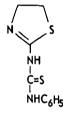
George W. A. Milne

Laboratory of Metabolism, National Heart Institute, National Institutes of Health, Bethesda, Maryland 20014 U.S.A.

(Received 3 November 1966)

Fromm and Kapeller-Adler (3) reported that the reaction of 2-amino-2-thiazoline with phenylisothiocyanate in ethanol when performed "under strong cooling" gave 2-imino-3-phenylthiocarbamoylthiazolidine (I), m.p. 60°,





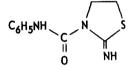
I II resolidification point 80° , m.p. 129°, and that the identical reactants, when heated on a steam bath, gave 2-phenylthiureido-2-thiazoline (II), m.p. 130° . Upon careful repetition of these experiments, only one mono-adduct, $C_{10}H_{11}N_3S_2$, m.p. $148-149^{\circ}$, was obtained as yellow crystals, regardless of the temperature at which the reaction was run.

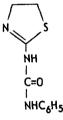
In order to distinguish between structures I and II for this product, an unequivocal synthesis of I was accomplished. The sodium salt of 2-aminoethanethiosulfuric acid was treated with phenylisothiocyanate as described by Ferris, Salerni and Schutz (4) and the resulting phenylthiureido derivative was converted in situ by the action of cyanide (1) in low yield to compound I, m.p. $148-149^{\circ}$. The infrared spectra of I made by the two

$$SO_3^{-2} + [C_6H_5NHCNHCH_2CH_2SCN] \longrightarrow 1$$

methods were identical and the mixture m.p. was undepressed.

The reaction of 2-amino-2-thiazoline with phenylisocyanate in acetonitrile gave a white crystalline mono-adduct, $C_{10}H_{11}N_3OS$, m.p. 149-150°, which was tentatively considered to be 2-imino-3-phenylcarbamoylthiazolidine (III), by analogy with the reaction between 2-amino-2-thiazoline and phenylisothiocyanate. An attempt was made to prepare an authentic sample





IV

Ш

of III, but while it was possible to make sodium 2-phenylureidoethylthiosulfate in acetonitrile solution, this compound failed to react with cyanide to give the desired cyclic product, only bis(2-phenylureidoethyl) disulfide, m.p. 198-203⁰, being isolated.

The conversion of I to III by the action of alkaline hydrogen peroxide according to the method of Papadopoulos (5) was similarly unsuccessful. The major product which was isolated was N,N'-diphenylurea, m.p. 240-241°, mixture m.p. 240-241°, which was identified by its infrared spectrum.

Behringer and Zilliken reported (6) that the Raney nickel degradation

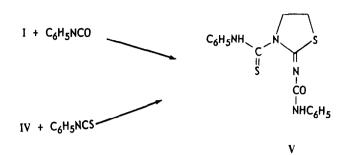
of 2-amino-2-thiazoline-4-carboxylic acid in aqueous solution gives α -alarine and it has been found (2) that similar treatment of 2-ureido-2-thiazoline yields urea. These results suggest that degradation of the thiazoline ring system occurs in the direction shown below:



Treatment of the phenylisocyanate adduct of 2-amino-2-thiazoline with Raney nickel gave N-phenylurea, m.p. $147-148^{\circ}$, mixture m.p. was undepressed. The infrared spectrum was identical with that of an authentic sample of N-phenylurea. This suggests that the phenylisocyanate adduct is 2-phenylure-ido-2-thiazoline (IV).

These findings were confirmed by a labelling experiment. Phenylisocyanate was condensed with $2-(^{15}N)$ -amino-2-thiazoline which had been prepared (2) from sodium 2-aminoethylthiosulfate and potassium cyanide- ^{15}N (99.7% isotopic purity), to give the phenylureido derivative. This adduct was treated with Raney nickel, as before, and the N-phenylurea which was isolated, m.p. 142-144°, was shown by mass spectrometry to contain the ^{15}N in the primary amino group. The molecular ion, $[C_7H_8^{14}N^{15}NO]^+$ requires m/e 137.061; found m/e 137.061. This result shows conclusively that the phenylisocyanate reacted with the exocyclic amino group. Cyanate ion has also been demonstrated (2) to react with the primary amino group of 2-amino-2-thiazoline.

Upon treating I with phenylisocyanate or IV with phenylisothiocyanate, the identical compound was obtained, $C_{17}H_{16}N_4S_2O$, m.p. 141.5-142.5^O, mixture m.p. was undepressed. These cross reactions, in which 2-phenylcarbamoylimino-3-phenylthiocarbamoylthiazolidine (V) was obtained, provide further evidence



for the correctness of the structural assignments of the two mono-adducts. All new compounds reported here gave satisfactory analytical data.

REFERENCES

 Paper I in this series: D. L. Klayman and G. W. A. Milne, J. Org. Chem., <u>31</u>, 2349 (1966).

2. Paper II in this series: D. L. Klayman, A. Senning and G. W. A. Milne, <u>Acta Chem. Scand.</u>, in press.

3. E. Fromm and R. Kapeller-Adler, <u>Ann.</u>, <u>467</u>, 240 (1928).

4. A. F. Ferris, O. L. Salerni and B. A. Schutz, <u>J. Chem. Soc</u>., 6650 (1965).

5. E. P. Papadopoulos, J. Org. Chem., 31, 3060 (1966).

6. H. Behringer and P. Zilliken, <u>Ann.</u>, <u>574</u>, 140 (1951).